

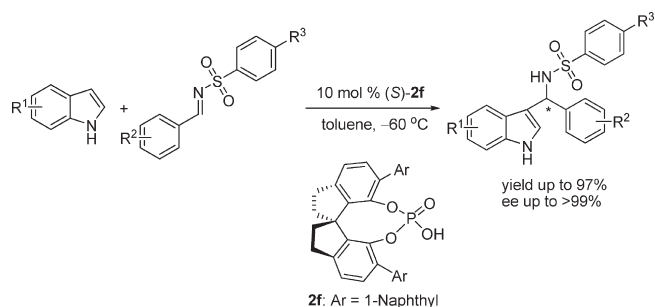
SPINOL-Derived Phosphoric Acids: Synthesis and Application in Enantioselective Friedel–Crafts Reaction of Indoles with Imines

Fangxi Xu,[†] Dan Huang,[†] Chao Han,[†] Wei Shen,[†] Xufeng Lin,^{*,†} and Yanguang Wang^{*,†,‡}

[†]Department of Chemistry, Zhejiang University, Hangzhou 310027, P. R. China, and [‡]State Key Laboratory of Applied Organic Chemistry, Lanzhou University, Lanzhou 730000, China

lxfoke@zju.edu.cn; orgwyg@zju.edu.cn

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A new class of chiral phosphoric acids with spirobiindane as scaffold were conveniently synthesized from (*S*)-1,1'-spirobiindane-7,7'-diol ((*S*)-SPINOL) and employed to catalyze the asymmetric Friedel–Crafts reaction of indoles with imines to afford 3-indolyl methanamines. High yields (68–97%) and excellent enantioselectivities (up to 99% ee) were obtained.

The catalytic asymmetric Friedel–Crafts (F–C) reaction is a powerful strategy for the construction of carbon–carbon bond in organic synthesis, providing direct approach to the enantio-merically enriched arene derivatives.¹ Since indoles exhibited significant nucleophilic reactivity and extensively biological activities,² their asymmetric Friedel–Crafts reactions are of great value.³ Among the published asymmetric Friedel–Crafts

reactions,^{4,5} the chiral phosphoric acids⁶ were demonstrated to be efficient.⁷ These phosphoric acid catalysts possess a biaryl backbone. Typical examples are BINOL and H₈-BINOL-derived phosphoric acids **1** (Figure 1). With regard to the significant works on the synthesis of enantiopure 1,1'-spirobiindane-7,7'-diol⁸ (SPINOL, **3**) and catalytic applications of its derivatives,⁹ we synthesized SPINOL-derived phosphoric acids **2** and investigated their application in enantioselective Friedel–Crafts reaction of indoles with imines, which furnished 3-indolyl methanamines. We herein report the results of this effort.

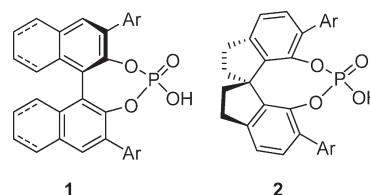


FIGURE 1. Chiral phosphoric acids.

As shown in Scheme 1, the SPINOL-derived phosphoric acids **2** were synthesized from (*S*)-SPINOL **3**. Initially, (*S*)-SPINOL was routinely protected with the MOM group to afford **4** in 94% yield. Compound **4** underwent lithiation and

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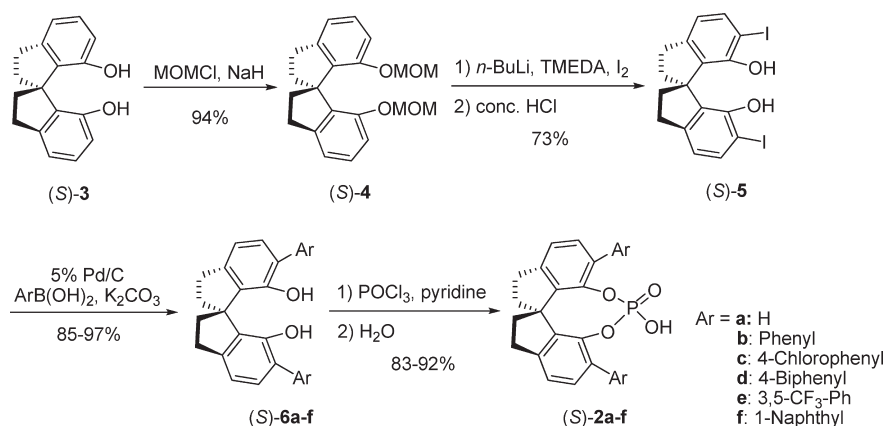
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SCHEME 1. Synthesis of SPINOL-Derived Phosphoric Acids 2



subsequent iodination, followed by deprotection of hydroxyl groups, to give 6,6'-diiodo compound **5** in 73% yield. Suzuki coupling reaction of **5** using 5% Pd/C as catalyst resulted in **6a–f** in excellent yields (85–97%) with convenient operation. Finally, phosphorylation of **6** afforded the corresponding phosphoric acids **2a–f** in high yields (83–92%).

TABLE 1. Optimization for Enantioselective Friedel–Crafts Reaction^a

entry	catalyst	temp	time	yield (%) ^b	ee (%) ^c
1	2a	rt	40 min	80	11
2	2b	rt	7 h	89	72
3	2c	rt	3 h	87	73
4	2d	rt	4 h	75	72
5	2e	rt	10 min	100	56
6	2f	rt	30 min	90	89
7	2f	0 °C	75 min	92	90
8	2f	−40 °C	8 h	85	94
9	2f	−60 °C	36 h	80	96

^aThe reaction was carried out with 0.5 mmol of **7a**, 0.1 mmol of **8a**, and 10 mol % of catalyst **2** in 0.5 mL toluene. ^bIsolated yield. ^cDetermined by HPLC analysis on a Chiralcel OD-H column.

To test the potential applications of our catalysts, the reaction of indole (**7a**) with *N*-tosylimine **8a** was chosen as a model reaction, which has been well promoted by chiral metal complex,^{4f,h} chiral thiourea,^{5b} and BINOL-derived phosphoric acid catalysts.^{7b} The products of this asymmetric Friedel–Crafts reaction provide easy access to the synthesis of enantiopure 3-indolyl methanamine derivatives. In our initial investigation, nonsubstituted **2a** was used as catalyst with 10 mol % loading, and the product was obtained in 80% yield with 11% ee (entry 1, Table 1), while BINOL phosphoric acid gave only racemic products.^{7b} Inspired by this result, we then tested the substituent effect at the 6,6'-positions of the catalysts with various aryl groups at room temperature and found the substituent effect to be remarkable (Table 1, entries 1–6). The reactions using **2b–d** as catalyst proceeded in similar yields and enantioselectivities, except that **2b** took a longer reaction time (Table 1, entries 2–4). When catalyst **2e** was employed, a quantitative yield as well as a moderate ee value (56%) were obtained within 10 min (Table 1, entry 5).

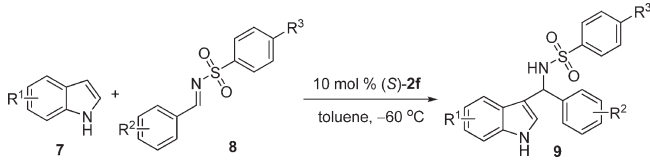
To our delight, the catalyst **2f**, bearing two 1-naphthyl groups at the 6,6'-positions of the SPINOL backbone, gave high yield (90%) and good enantioselectivity (89% ee) at room temperature. Further improvement of the enantioselectivity was achieved at lower reaction temperature (Table 1, entries 7–9). Up to 96% ee enantioselectivity and 80% yield were obtained when the reaction was performed at −60 °C for 36 h (Table 1, entry 9).

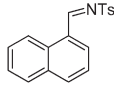
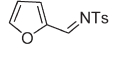
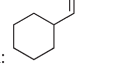
The substrate scope of the enantioselective Friedel–Crafts reaction of indoles **7** with imines **8** was then evaluated and is summarized in Table 2. Interestingly, both the reactivity and the enantioselectivity were enhanced by changing the tosyl group of *N*-sulfonylphenylimines to the more electron-withdrawing *p*-bromobenzenesulfonyl group (Table 2, entries 1–3). Indoles bearing different substituents were then examined for the reaction with imines **8a** or **8c**, and the result showed a good tolerance in this catalytic system (Table 2, entries 4–9), although the introduction of an electron-withdrawing group onto the indole ring led to a lower reactivity (Table 2, entries 8 and 9). For aryl imines **8d–i**, the reactions went smoothly to give the corresponding products in 68–93% yield and 91–>99% ee (Table 2, entries 10–16). In the case of **8i** derived from 1-naphthaldehyde, excellent enantioselectivity (97% ee) was obtained at higher temperature (0 °C) (Table 2, entry 16). In addition, heteroaryl imine **8j** also gave excellent yield (96%) and enantioselectivity (97% ee) (Table 2, entry 17). For alkyl imine **8k**, however, we obtained the adduct in lower yield (47%) and moderate enantioselectivity (79% ee) (Table 2, entry 18). The *S*-configuration adduct product **9ad** was confirmed by the X-ray crystal structure analysis (see Supporting Information).

Recently, Simon and Goodman¹⁰ provided both data and computation models to identify the specific indolyl methanamine product for Friedel–Crafts reactions of indole with *N*-tosylimines catalyzed by BINOL-phosphoric acids. On the basis of this result, we propose a possible model for the asymmetric induction of our catalytic system as shown in Scheme 2. Phosphoric acid (**S**)-**2** as a bifunctional catalyst combines two substrates through hydrogen bonding. In this model, indole attacks *N*-tosylimine from the *re* face preferentially, leading to a *S*-configuration adduct.

In conclusion, we have synthesized a new class of chiral phosphoric acids with a spirobiindane backbone, which could efficiently promote the enantioselective Friedel–Crafts reaction of indoles with imines to afford 3-indolyl methanamines. Up

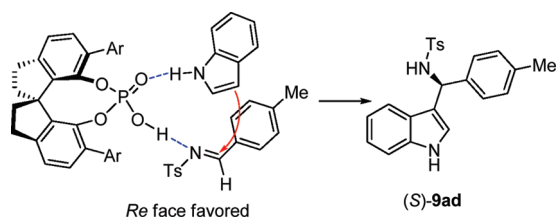
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TABLE 2. Enantioselective Friedel–Crafts Reaction Catalyzed by **2f**^a


entry	7 (R ¹)	8 (R ² /R ³)	time (h)	product	yield (%) ^b	ee (%) ^c
1	7a (H)	8a (H/Me)	36	9aa	80	96
2	7a	8b (H/H)	24	9ab	85	96
3	7a	8c (H/Br)	18	9ac	89	99
4	7b (2-Me)	8a	12	9ba	97	91
5	7b	8c	8	9bc	93	93
6	7c (5-OMe)	8a	24	9ca	85	97
7	7c	8c	28	9cc	81	99
8 ^d	7d (5-Br)	8a	36	9da	85	93
9 ^d	7d	8c	24	9dc	79	94
10	7d	8d (4-Me/Me)	24	9dd	81	>99
11	7a	8d	20	9ad	92	>99(S) ^f
12	7a	8e (3,4-Me ₂ /Me)	40	9ae	77	98
13	7a	8f (4-OMe/Me)	48	9af	76	97
14	7a	8g (4-Cl/Me)	60	9ag	68	91
15	7a	8h (3-OMe/Br)	26	9ah	93	97
16 ^e	7a	8i : 	12	9ai	85	97
17	7a	8j : 	9	9aj	96	97
18	7a	8k : 	10	9ak	47	79

^aUnless otherwise noted, reactions were carried out with 0.5 mmol of **7**, 0.1 mmol of **8**, and 10 mol % of **2f** in 0.5 mL of toluene at $-60\text{ }^{\circ}\text{C}$. ^bIsolated yield. ^cDetermined by HPLC analysis on a Chiralcel OD-H column. ^dReaction at $-40\text{ }^{\circ}\text{C}$. ^eReaction at $0\text{ }^{\circ}\text{C}$. ^fAbsolute configuration determined by X-ray single-crystal analysis.

SCHEME 2. Proposed Reaction Model



to 97% yield and up to 99% ee enantioselectivity were obtained. Further catalytic applications of the SPINOL-derived phosphoric acids in other enantioselective reactions are in progress.

Experimental Section

Synthesis of (S)-6,6'-Diiodo-1,1'-spirobiindane-7,7'-diol (5). To a solution of (S)-**4**^{9d} (0.81 g, 2.4 mmol) and TMEDA (0.75 mL, 5 mmol) in Et₂O (15 mL) was added *n*-BuLi (2.5 M in hexane, 2.88 mL, 7.2 mmol) at $-78\text{ }^{\circ}\text{C}$. After being stirred at room temperature for 6 h, the mixture was cooled to $-78\text{ }^{\circ}\text{C}$, and a solution of

iodine (1.83 g, 7.2 mmol) in Et₂O (20 mL) was carefully added. The resulted suspension was warmed to room temperature, stirred overnight, quenched by saturated Na₂SO₃, and stirred for an additional 1 h. The organic layer was separated, and the aqueous layer was extracted by ether. The combined organic layer was washed with brine, dried over Na₂SO₄, passed through a pad of silica, and concentrated in vacuo, giving the crude (S)-6,6'-diiodo-7,7'-bis(1-methoxymethoxy)-1,1'-spirobiindane as brown solid, which was used without further purification in the next step. The crude (S)-6,6'-diiodo-7,7'-bis(1-methoxymethoxy)-1,1'-spirobiindane was dissolved in CHCl₃ (12 mL) and MeOH (18 mL), and conc HCl (12 mL) was added. After being refluxed for 3 h, the mixture was poured into water and extracted by CH₂Cl₂. The organic layer was washed with saturated NaHCO₃, dried over Na₂SO₄, and concentrated. The residue was purified by silica gel column chromatography (ethyl acetate/petroleum ether = 1:10) to give the product **5** (0.88 g, 73% yield) as white solid. Mp 149–150 °C; $[\alpha]_{\text{D}}^{20} = -187.2$ (c 1.0, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 2.18–2.24 (m, 2 H), 2.29–2.37 (m, 2 H), 2.94–3.06 (m, 4 H), 5.11 (s, 2H), 6.65 (d, *J* = 8.0 Hz, 2 H), 7.52 (d, *J* = 7.6 Hz, 2 H); ¹³C NMR (100 MHz, CDCl₃) δ 31.1, 37.7, 59.8, 83.2, 119.2, 133.0, 137.7, 146.7, 151.1; ATR-FTIR 3479, 2935, 1571, 1436, 1323, 1290, 1233, 1188, 1061, 1000, 796,

754 cm^{-1} ; HRMS (ESI) calcd for $\text{C}_{17}\text{H}_{13}\text{I}_2\text{O}_2^-$ ($[\text{M} - \text{H}]^-$): 502.9005. Found: 502.8989.

Typical Synthesis of (*S*)-6,6'-Diphenyl-1,1'-spirobiindane-7,7'-diol (6b**).** A solution of (*S*)-**5** (151 mg, 0.3 mmol), phenylboronic acid (128 mg, 1.05 mmol), K_2CO_3 (0.145 g, 1.05 mmol), and 5% Pd/C (13 mg; 0.006 mmol) in dioxane/water (1:1, 6 mL) was stirred at 80 °C for 2 h and then diluted with ethyl acetate and 3 N HCl. The organic layer was washed with brine, dried over Na_2SO_4 , and concentrated in vacuo. The residue was purified by silica gel column chromatography (ethyl acetate/petroleum ether = 1:15) to give the product **6b** (117 mg, 97% yield) as white solid. Mp 203–204 °C; $[\alpha]_{\text{D}}^{20} = -281.5$ (*c* 0.9, CHCl_3); ^1H NMR (400 MHz, CDCl_3) δ 2.34–2.46 (m, 4 H), 3.02–3.15 (m, 4 H), 5.07 (s, 2H), 6.93 (d, *J* = 7.6 Hz, 2 H), 7.19 (d, *J* = 7.6 Hz, 2 H), 7.30 (t, *J* = 7.2 Hz, 2 H), 7.39 (t, *J* = 7.2 Hz, 4 H), 7.47 (d, *J* = 7.6 Hz, 4 H); ^{13}C NMR (100 MHz, CDCl_3) δ 31.1, 37.7, 58.4, 117.4, 126.9, 127.2, 128.6, 129.3, 130.5, 132.0, 137.4, 145.2, 149.4; ATR-FTIR 3510, 2943, 1616, 1580, 1499, 1472, 1420, 1253, 1228, 1180, 1074, 998, 814, 753, 698 cm^{-1} ; HRMS (ESI) calcd for $\text{C}_{29}\text{H}_{23}\text{O}_2^-$ ($[\text{M} - \text{H}]^-$): 403.1698. Found: 403.1683.

Typical Synthesis of (*S*)-1,1'-Spirobiindane-7,7'-diyl phosphate (2f**).** To a solution of (*S*)-**6f** (252 mg, 0.5 mmol) in pyridine (3 mL) was added POCl_3 (92 μL , 1.0 mmol) dropwise at 0 °C, and the resulting mixture was heated to 70 °C and stirred for 3 h. After the mixture cooled to 0 °C, 3 mL of water was carefully added, and the resulting suspension was stirred at 110 °C for an additional 4 h. Dichloromethane was added, and the pyridine was removed by reverse extraction with 4 N HCl. The organic layer was dried over Na_2SO_4 and concentrated. The residue was purified by silica gel column chromatography (dichloromethane/methanol = 1:20 to 1:10) to give the product **2f** (257 mg, 91% yield) as a foamy solid. Mp > 300 °C; $[\alpha]_{\text{D}}^{20} = -362.2$ (*c* 0.50, CHCl_3); ^1H NMR (400 MHz, CHCl_3) δ 2.34–2.51 (m, 4H), 2.90–2.97 (m, 2H), 3.19–3.25 (m, 2H), 6.95–7.02 (m, 2H), 7.19–7.25 (m, 6H), 7.36–7.49 (m, 4H), 7.56–7.61 (m, 2H), 7.70–7.81 (m, 4H); ^{13}C NMR (100 MHz, CHCl_3) δ 30.3, 39.1, 59.9, 120.7, 121.9, 125.2, 125.8, 126.5, 127.8, 129.8, 131.3, 131.7, 132.5, 132.7, 133.4, 137.4, 140.5, 144.4, 145.6; ^{31}P NMR (202 MHz, CDCl_3) δ -9.9; ATR-FTIR 3668, 2982,

2902, 1612, 1398, 1249, 1059, 871, 802, 777 cm^{-1} ; HRMS (ESI) calcd for $\text{C}_{37}\text{H}_{26}\text{O}_4\text{P}^-$ ($[\text{M} - \text{H}]^-$): 565.1569. Found: 565.1541.

Typical Procedure for the Enantioselective Friedel–Crafts Reaction. To a mixture of *N*-sulfonyl imine **8** (0.1 mmol) and catalyst **2f** (0.01 mmol) was added toluene (0.5 mL). Then the mixture was stirred for 10 min at room temperature. Indole **7** (0.5 mmol) was subsequently added in one portion at -60 °C. After the reaction was complete, it was quenched by 2 M NaOH and extracted with ethyl acetate. The organic layer was washed with brine, dried over Na_2SO_4 , and concentrated in vacuo. The residue was purified by silica gel column chromatography (ethyl acetate/petroleum ether = 1:3) to give the corresponding product **9**. Product **9ac** was obtained in 89% yield after chromatography and 99% ee as determined by HPLC [Daicel Chiralcel OD-H, *n*-hexane/*i*-propanol = 70:30, 0.8 mL/min, $\lambda = 254$ nm, t_{R} (minor) = 15.69 min, t_{R} (major) = 35.37 min]. Mp 135–136 °C; $[\alpha]_{\text{D}}^{20} = -88.3$ (*c* 0.80, Acetone); ^1H NMR (400 MHz, CDCl_3) δ 5.24 (d, *J* = 7.2 Hz, 1 H), 5.91 (d, *J* = 7.2 Hz, 1H), 6.65 (s, 1 H), 7.06 (t, *J* = 7.6 Hz, 1 H), 7.18–7.22 (m, 6 H), 7.30–7.38 (m, 4 H), 7.42 (d, *J* = 8.0 Hz, 2 H), 7.99 (br, 1 H); ^{13}C NMR (100 MHz, CDCl_3) δ 55.2, 111.4, 115.8, 119.3, 120.1, 122.6, 123.8, 125.3, 126.9, 127.2, 127.5, 128.38, 128.44, 131.6, 136.5, 139.4, 139.6; ATR-FTIR 3669, 3411, 2982, 2902, 1399, 1323, 1249, 1157, 1071, 896, 740 cm^{-1} ; HRMS (ESI) calcd for $\text{C}_{21}\text{H}_{17}\text{BrN}_2\text{O}_2\text{SNa}^+$ ($\text{M} + \text{Na}^+$): 463.0092. Found: 463.0072.

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Supporting Information Available: Detailed experimental procedures, characterization data, CIF file, and copies of ^1H and ^{13}C NMR and HPLC spectra. This material is available free of charge via the Internet at <http://pubs.acs.org>.